

# Chapter 19 Study Guide Acids Bases Answers

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### Chapter 19 Study Guide Acids

Chapters 19 and 20 Study Guide Chapter 19 Vocabulary: Lewis Acid-any substance that can accept a pair of electrons to form a covalent bond Lewis Base-any substance that can donate a pair of electrons to form a covalent bond Acid - a compound that produces hydrogen ions in solution

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compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts.

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taste bitter, feel slippery, will change the color of an acid-base indicator, and can be strong or weak electrolytes in aqueous solution.

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