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HOLT MODERN CHEMISTRY -- CHAPTER 1 REVIEW The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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1: Chemistry Is a Physical Science: Section 1 Review: p.5: 2: Matter and Its Properties: Section 2 Review: p.14: 3: Elements: Section 3 Review: p.20: Chapter Review: p.22

Solutions to Modern Chemistry (9780030367861) :: Homework ...

CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

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1. a Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not involve chance discoveries. (c) is driven by curiosity. (d) is done for the sake of learning something new.

1 Matter and Change - HUBBARD'S CHEMISTRY

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided.

1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$

4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O_2 :1 mol H_2O c. What is the mole ratio ...

mc06se cFMsr i-vi

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to

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decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H_2SO_3 6 b. S in MgSO_4 2 c. S in K_2S 1 d. Cu in Cu_2S 6 e. Cr in Na_2CrO_4 5 f. N in HNO_3 4 g. C in HCO_3^- ...

7 Chemical Formulas and Chemical Compounds

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SECTION 1.1 REVIEW VOCABULARY 1. Define chemistry. 2. What types of substances do chemists work with? REVIEW 3.

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each kind in a chemical compound. Reveals the number of
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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

3 Atoms: The Building Blocks of Matter

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1: Review of Chemical Bonding ... Group essay project: In-class

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discussion questions from this chapter: Group essay project: Join a group of classmates from your inorganic chemistry course. ... the quantum mechanical theory of atomic orbitals and is considered one of the most important papers in the history of modern chemistry. At the meeting ...

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